| Activity 14 – 2: Glass Evidence – Density and Refractive Index  |
|---|
| <b>Background:</b> In this activity, you will be asked to determine the density and refractive index of glass found at the crime scene and the density and refractive indices of glass found on suspects. If these two characteristics do not match, you might be able to disqualify a suspect. Note glass is class evidence. |
| Purpose:  |
|   |

Period: \_\_\_\_\_ Date: \_\_\_\_\_

#### Procedure:

### Part 1. Observation of Glasses

Observe the glasses collected from each suspect. Write your observation.

### Part 2. Density of Glasses

- a. Take the mass of one type of glass beads using the electronic balance. Record the mass.
- b. Place small amount of water in the graduated cylinder about 3mL (initial volume).
- c. Place all of the same type beads in the graduated cylinder and read the volume (final volume). Determine final volume by subtrating initial from final volume. Record.
- d. Divide mass by volume to obtain density of each glass type of glass.

#### Part 3. Refractive Index of Glasses

- a. Pour the refractive index solution #1 into the cup from the vial.
- b. Place one type of glass beads. Write your observation (visible appears larger or smaller, disappers).
- c. Pour the refractive index solution #1 back into the vial.
- d. Place the glass beads on a paper towel, clean, and place back in the vial.
- e. Repeat the procedure for second and third types of beads.
- f. Repeat steps a e for Refractive Index solution #2 and #3 for all glass beads.

### Data:

Table 1. Observation of Glass

|                | Type of Glass | Observation |
|----------------|---------------|-------------|
| Suspect 1 (S1) | Borosilicate  |             |
| Suspect 2 (S2) | Flint         |             |
| Suspect 3 (S3) | Soda-lime     |             |
| Crime Scene    |               |             |

# Table 2. Density of Glass

| Sample          | Mass (g) | Volume (mL) | Density (g/mL) |
|-----------------|----------|-------------|----------------|
| S1 Borosilicate |          |             |                |
| S2 Flint        |          |             |                |
| S3 Soda-lime    |          |             |                |
| Crime Scene     |          |             |                |

## Table 3. Refractive Index

|                 | Refractive Index<br>Solution #1 | Refractive Index<br>Solution #2 | Refractive Index<br>Solution #3 |
|-----------------|---------------------------------|---------------------------------|---------------------------------|
| S1 Borosilicate |                                 |                                 |                                 |
| S2 Flint        |                                 |                                 |                                 |
| S3 Soda-lime    |                                 |                                 |                                 |
| Crime Scene     |                                 |                                 |                                 |

# Analysis:

- 1. Did the density of the glass found on any of the four suspects match the density of the glass found at the crime scene? Explain your answer.
- 2. Explain why glass is considered a class evidence.

Conclusion: